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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.009 \AA$
Disorder in main residue
$R$ factor $=0.062$
$w R$ factor $=0.166$
Data-to-parameter ratio $=11.7$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## ( $\boldsymbol{\eta}^{5}$-Cyclopentadienyl) $\boldsymbol{\eta}^{4}$-(exo-5-phenyl-ethynyl)-1,3-cyclopentadienyl]cobalt(II)

In the crystal structure of the title compound, $\left[\mathrm{Co}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\right.$ $\left(\mathrm{C}_{13} \mathrm{H}_{10}\right)$ ], the Co atom is connected to two five-membered rings. Both rings are disordered and were refined using a split model.

## Comment

The cyclopentadienyl anion, $\mathrm{C}_{5} \mathrm{H}_{5}(\mathrm{Cp})$, is one of the most important ligands in the organometallic chemistry of transition metals (Jutzi et al., 1994; Janiak et al., 1991). Such ligands can stabilize metals in low and high oxidition states, while the easy change of hapticity ( $\eta^{1}, \eta^{3}$ or $\eta^{5}$ ) of the Cp ligand allows it to be readily adaptable to changes in the electronic and steric requirements of the central atom (Philippopoulos et al.,1998). In this context, we have focused our attention on the preparation of cobaltocenium derivatives. The title compound, (I), is a synthetic intermediate from which different cobaltocenium derivatives with e.g. good solubility in polar solvents can be prepared.

(I)

In the structure of the title compound, both five-membered rings are disordered and were refined using a split model (Fig. 1). Atom C10 deviates significantly from the $\eta^{4}\left(\mathrm{C}_{4} \mathrm{H}_{4}\right)$ ring plane; this is obvious from the torsion angles $\mathrm{C} 10-\mathrm{C} 6-$ $\mathrm{C} 7-\mathrm{C} 8, \mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ and $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 6$ (Table 1). Therefore, atoms C6, C7, C8, C9, C10 are not coplanar. The dihedral angle between the $\eta^{4}\left(\mathrm{C}_{4} \mathrm{H}_{4}\right)$ ring plane and the plane formed by C6, C9 and C10 is $29.4(1)^{\circ}$.

## Experimental

The title compound was synthesized according to literature procedures (Wildschek et al.,1990). Crystals appropriate for data collection were obtained by slow evaporation of a mixed dichloromethane and hexane solution ( $2: 1 \mathrm{v} / \mathrm{v}$ ) at room temperature.

## Crystal data

$\left[\mathrm{Co}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mathrm{C}_{13} \mathrm{H}_{10}\right)\right]$
$M_{r}=290.23$
Monoclinic, $P 2_{1} / c$
$a=6.0403(7) \AA$
$b=17.926(2) \AA$
$c=12.9002(14) \AA$
$\beta=92.134(2)^{\circ}$
$V=1395.9(3) \AA^{3}$
$Z=4$
$D_{x}=1.381 \mathrm{Mg} \mathrm{m}^{-3}$
[ $\mathrm{Co}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mathrm{C}_{13} \mathrm{H}_{10}\right)$ ]
$r=290.23$
$a=6.0403$ (7) A
$b=17.926$ (2) A
$\beta=92.134$ (2) ${ }^{\circ}$
$V=1395.9$ (3) $\AA^{3}$
$Z=4$

## Mo $K \alpha$ radiation

Cell parameters from 879
reflections
$\theta=2.3-16.9^{\circ}$
$\mu=1.21 \mathrm{~mm}^{-1}$
$T=292$ (2) K
Needle, red
$0.40 \times 0.10 \times 0.06 \mathrm{~mm}$

## Data collection

Bruker SMART APEX CCD area-
detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: none
6946 measured reflections
2443 independent reflections
1450 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.056$
$\theta_{\text {max }}=25.0^{\circ}$
$h=-7 \rightarrow 7$
$k=-21 \rightarrow 18$
$l=-13 \rightarrow 15$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.062$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0697 P)^{2}\right. \\
& \quad+0.4218 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.00 \\
& \Delta \rho_{\max }=0.45 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.38 \mathrm{e} \AA^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| C10-C11 | $1.499(7)$ | $\mathrm{C} 12-\mathrm{C} 13$ | $1.435(7)$ |
| :--- | :---: | :--- | ---: |
| $\mathrm{C} 11-\mathrm{C} 12$ | $1.178(7)$ |  |  |
| $\mathrm{C} 7-\mathrm{C} 6-\mathrm{C} 10$ | $112.2(8)$ | $\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 6$ | $89.3(6)$ |
| $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $116.9(8)$ |  |  |
| $\mathrm{C} 10-\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $-21.6(10)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $23.6(9)$ |
| $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ | $-0.6(9)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 6$ | $-32.7(8)$ |

It was apparent at an early stage of the refinement that both fivemembered rings were disordered. They were therefore refined using a split model with DFIX restraints (SHELXS97; Sheldrick, 1997) of $\mathrm{C}=\mathrm{C}=1.40$ (1) $\AA, \mathrm{C}-\mathrm{C}=1.52$ (1) $\AA$ for the $\mathrm{C}-\mathrm{C}$ distances. The major component was refined anisotropically whereas the minor component was refined isotropically. However, for some atoms the disorder could not be resolved and therefore they show enlarged anisotropic displacement parameters. The site-occupation factors for the disordered atoms were refined to 0.58 (4) and 0.42 (4), respec-


Figure 1
The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the $30 \%$ probablity level. Both disorder components of each ring are shown.
tively, for the major and minor components of the cyclopentadienyl ring, and to 0.64 (3) and 0.36 (3), respectively, for the major and minor components of atoms C6-C8. H atoms were positioned geometrically, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-0.98 \AA$, and refined as riding, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: SMART (Bruker, 2001); cell refinement: SMART; data reduction: SAINT (Bruker, 2001); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL.

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